From Docking False-Positive to Active Anti-HIV Agent

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Virtual screening of the Maybridge library of ca. 70 000 compounds was performed using a similarity filter, docking, and molecular mechanics—generalized Born/surface area postprocessing to seek potential non-nucleoside inhibitors of human immunodeficiency virus-1 (HIV-1) reverse transcriptase (NNRTIs). Although known NNRTIs were retrieved well, purchase and assaying of representative, top-scoring compounds from the library failed to yield any active anti-HIV agents. However, the highest-ranked library compound, oxadiazole 1, was pursued as a potential "near-miss" with the BOMB program to seek constructive modifications. Subsequent synthesis and assaying of several polychloro-analogs did yield anti-HIV agents with EC_{50} values as low as 310 nM. The study demonstrates that it is possible to learn from a formally unsuccessful virtual-screening exercise and, with the aid of computational analyses, to efficiently evolve a false positive into a true active.

Introduction

The acquired immunodeficiency syndrome (AIDS) crisis continues with ca. 40 million people infected by human immunodeficiency virus (HIV) and 2.9 million HIV-related deaths in 2006.1 The virally encoded proteins of HIV provide chemotherapeutic targets for the treatment of infection by the virus. A principal point of attack has been HIV reverse transcriptase (RT), which is required for the conversion of the viral genomic RNA to DNA and for successful infection of host cells. This has led to the development and FDA approval of two important classes of anti-HIV drugs: (i) nucleoside and nucleotide RT inhibitors (NRTIs and NtRTIs, respectively), for example, AZT, ddI, ddC, d4T, FTC, TDF, and 3TC, and (ii) non-nucleoside RT inhibitors (NNRTIs), specifically, nevirapine, delavirdine, and efavirenz. The NRTIs and NtRTIs are actually faulty substrates that cause premature termination of the growing DNA transcript, while NNRTIs are true inhibitors that bind to an allosteric pocket in the vicinity of the polymerase active site.² The therapeutic situation is challenged by rapid mutation of the virus to yield resistant strains. This leads to a need for new drugs with activity against at least parts of the spectrum of variants, which are now clinically observed.

Our own efforts have been directed toward the development of NNRTIs with enhanced therapeutic spectra and auspicious pharmacological properties. The approach to date has featured focused synthetic organic chemistry and anti-HIV assaying driven by automated procedures for the creation and evaluation of virtual libraries, estimation of pharmacological properties, and lead optimization featuring free-energy perturbation calculations to assess relative protein—ligand binding affinities.³ Highly potent and structurally diverse anti-HIV agents have been discovered;³ however, we continue to seek activity against an ever-wider range of viral mutants and exploration of alternative structural classes for NNRTIs. To this end, reported successes for lead generation by molecular docking have been intriguing,⁴ and it was decided to try this approach to seek novel NNRTIs. The following report provides a case study on a common dilemma in a virtual or high-throughput screening (HTS) exercise. It is demonstrated that, with confidence in computed structures and estimated activities, it is possible to convert a false positive into an active agent.

Virtual Screening

A summary of the virtual screening effort is presented here, with full details provided elsewhere.5 Virtual screening was performed on roughly 70 000 compounds from the Maybridge collection, which was supplemented with 26 known NNRTIs. The screening protocol began with a similarity filter that retrieves 60% of the known actives in the top 5% of the screened library. The ca. 2000 library compounds that were most similar to the known actives were then docked into the structure of wild-type HIV-1 RT, taken from its 1rt4 crystal structure⁶ in complex with a carboxanilide NNRTI, using Glide 3.5 with standard precision.7 The top 500 compounds were then redocked and scored in Glide extra-precision (XP) mode. The top 100 of these were post-scored with a molecular mechanicsgeneralized Born/surface area (MM-GB/SA) method that was shown to provide high correlation between predicted and observed activities for known NNRTIs.5 The MM-GB/SA procedure included a conformational search for the ligand and optimization of the complex with the all-atom optimized potentials for liquid simulations (OPLS-AA) force field⁸ as well as evaluation of the change in free energy of hydration using the GB/SA approach,9 as implemented in MacroModel.10

The final top-20 ranked compounds consisted of 10 library compounds and 10 known NNRTIs. The best ranked library compound, **1** (S10087), was third overall, and the other top-20 library compounds were in two series containing a dihydrobenzofuran or 1-phenyltetrazole. The known NNRTIs are **2** (SDABO-3w), **3** (UC781), **4** (TMC125), **5** (efavirenz) and three analogs, and three pyrimidine derivatives from our group.³ Six of the top library compounds were purchased from Maybridge:

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1 and the KM and NRB compounds shown above. The two NRB compounds were found to be impure and gave mass spectra that are inconsistent with the illustrated structures. The other four library compounds gave correct NMR and mass spectra and were purified by high-performance liquid chromatography, recrystallized, and submitted for assay. Activities against the IIIB strain of HIV-1 were determined using MT-2 human T-cells; EC_{50} values are obtained as the dose required to achieve 50% protection of the infected cells using the MTT colorimetric method. The CC_{50} for inhibition of MT-2 cell growth by 50% is obtained simultaneously.¹¹ The four compounds showed no anti-HIV activity up to their CC_{50} concentrations of 61, 41, 75, and 3 μ M, respectively.



In essence, our best effort was made to computationally screen the Maybridge collection in search of NNRTI leads. The screening protocol seemed well-founded in view of (a) the testing of the MM-GB/SA procedure, (b) the fact that Glidedocked structures ("poses") for nine known NNRTIs agreed well with crystallographic results, and (c) the fact that the NRB compounds are close in structure to recently reported NNRTI HTS-hits that feature *N*-phenyl triazoles and tetrazoles (Scheme 1).¹²

In fact, 16 compounds had been purchased from Maybridge and assayed from a preliminary version of the same docking exercise that did not include the MM-GB/SA post-scoring. These compounds were also all inactive in the MT-2 assay, and structure visualization indicated that there were generosities in the evaluation of conformational energetics by Glide, including some twisted groups such as amides that should be more-orless planar. This led to exploration of the post-scoring and to the reporting of an analysis of the current problems for rankordering of compounds by docking owing to the uncertainties in evaluating conformational energetics and "conformer focusing", basically the need to evaluate the overall change in conformational energetics for the ligand and protein between the unbound and bound states.¹³ In the end, the structures from the docking plus MM-energy minimization appear reasonable,



Figure 1. Computed structure for 1 bound to HIV-1 RT from Glide docking followed by energy minimization with *MacroModel*. Only protein residues in the first layer around the ligand are illustrated. The ligand is positioned as expected from known structures such as for 4.^{16,17}

as illustrated in Figure 1 for **1**, and the penalties for achieving the docked conformations were not high.

Post Mortem. Several reasons can be suggested for the lack of success with finding active compounds in the Maybridge library: (a) remaining inaccuracies in the scoring including the "conformer focusing" problem and lack of explicit representation of the solvent, (b) the size of the screened library (only 70 000 compounds), (c) needing to dig deeper into the deck (purchase more compounds), and (d) the sensitivity of activity to structure. Concerning reason d, as demonstrated in our own NNRTI series, the addition of a chloro and a methoxy group takes 6 from an EC_{50} of 30 μ M to 10 nM (8),³ while the addition of another small group in the wrong place can be very detrimental (9 or **10**).



Thus, it is expected that there can be "near-misses" from virtual or experimental screening in which the activity window for an assay is missed owing to a seemingly minor structural omission or addition. Even given a viable core, the chance of an active substituent pattern for a target being included in a molecular library is extremely low.¹⁴ Of course, the nature of the assay can also be an issue. It is possible that some inactive compounds in a cell-based assay might be active in an in vitro assay for enzyme inhibition. However, in the present case, good correlations have been observed consistently for NNRTIs in



Figure 2. Training results for scoring using *BOMB* for 339 complexes with HIV-RT (magenta: uracil derivatives; green: analogs of 2), COX-2 (blue), FKBP (red), and p38 kinase (black). The correlation coefficient for the fit, r^2 , is 0.58, and the root-mean-square error is 0.82 log unit. Indicator variables are used to provide constant offsets for three datasets.

assays for the inhibition of reverse transcription and in cellbased anti-HIV analyses.¹⁵ Furthermore, the value of a lead that was not active in the cell-based assay would be comparatively low.

From False-Positive to Lead

The possibility that the virtual screening provided a nearmiss was pursued to ascertain whether a false-positive could be evolved into a lead and to determine how close the miss was in structure to a true active. The oxadiazole motif was chosen since 1 ranked so highly in the virtual screening. The predicted structure of the complex (Figure 1) seemed reasonable and reminiscent of the structures of complexes of HIV-RT with analogs of 4.^{16,17} Three analogs of 1 differing in the substitution of the anilinyl ring were also in the top-50 ranked compounds: 3-fluoro (S10076), 3-chloro, 4-methyl (S10085), and 2,4dichloro (S10089). All four compounds share the 3,4-dimethoxybenzyl group. Suspicion centered on this subunit and on the oxadiazole ring itself. 1 and 4 can be viewed as members of a general structural class, U-Het-NH-PhX,3a where U is an unsaturated group such as a substituted benzyl, anilinyl, or phenoxy group, Het is a heterocycle, and PhX is a substituted phenyl group.¹⁸ NNRTIs have been reported by Janssen and co-workers where Het is a pyrimidine, triazine, or pyrazinone.^{19,20} However, there have not been reports where U is a 3,4-disubstituted benzyl group or Het is a 5-membered-ring heterocycle. Finally, the para-methyl substituent in 1 was not viewed as optimal; chloro and cyano are common in the Janssen NNRTIs, and, in our series, 4-chloro and cyano are preferred over methyl by a factor of ca. 15 in EC₅₀ values.³ Thus, surgery was indicated for both phenyl rings of 1, though it was possible that no oxadiazole derivative would be viable.

BOMB Analyses. At this point, a full substituent scan was indicated for replacement of each phenyl hydrogen in the 2-anilinyl-5-benzyloxadiazole core. This was carried out with the BOMB program, which creates analogs by adding substituents to a core that has been placed in a binding site.^{3a} A thorough conformational search is performed for each analog, and the position, orientation, and dihedral angles for the analog are optimized using the OPLS-AA force field for the protein and OPLS/CM1A for the analog.⁸ The protein is rigid except for optimization of the terminal dihedral angles for side chains with hydrogen-bonding groups (e.g., the OH of tyrosine and the carboxylate group of aspartate). The resultant conformer for each analog with the lowest energy is then evaluated with



Figure 3. Structure for the complex of 1 with HIV-RT as computed by *BOMB*. The 176 residues nearest the ligand are included in the calculations.

a docking-like scoring function. The current scoring function has been trained to reproduce experimental activity data for 339 complexes of HIV-RT, COX-2, FK506 binding protein (FKBP), and p38 kinase (Figure 2). The scoring function only contains five descriptors that were obtained by linear regression: the octanol/water partition coefficient for the analog as computed by QikProp (QPlogP_{0/w}),²¹ the amount of hydrophobic surface area for the protein that is buried upon complex formation $(\Delta FOSA_P)$, the number of potential hydrogen-bond donating hydrogens in the analog (HBDN_L), the number of nonconjugated amides in the analog, and the number of "bad" protein-analog contacts in the complex (NBAD). The latter represent structural mismatches between two atoms within 4 Å, typically between a potential hydrogen-bonding oxygen or nitrogen and a saturated carbon atom or between a potential hydrogen-bond accepting O or N and another such atom or an aryl carbon atom.

Interestingly, (a) the most significant descriptor is $\text{QPlogP}_{o/w}$, which alone yields a fit with an r^2 of 0.47, and (b) inclusion of energetic results from full conjugate-gradient optimizations of the complexes created by BOMB was found to have little impact on the accuracy of the scoring. Although the BOMB scoring is still under development, the current version provides a useful evaluation of potential activity (Figure 2).

A standard protocol for a substituent scan with BOMB is to replace each hydrogen of a core with 10 small groups that have been selected for difference in size, electronic character, and hydrogen-bonding patterns: Cl, CH₃, NH₂, OH, CH₂NH₂, CH₂-OH, CHO, CN, NHCH₃, and OCH₃. To begin, the structure of 1 bound to RT was built with BOMB using the anilinylbenzyloxadiazole core and the structure of RT from the 1s9e PDB file.²² The switch from the 1rt4 structure arose because the 1s9e structure has higher resolution (2.6 vs 2.9 Å), and the 1s9e ligand is more similar to the U-Het-NH-PhX motif of the oxadiazole: it is a U-Het1-L-Het2 member with Het1 as a pyridinone, U as 3,5-dimethylphenoxy, and L, a linking chain, as CH₂SCH₂. The structure of the 1 complex built using BOMB is illustrated in Figure 3. It is similar to the structure in Figure 1 except for the conformation of the two methoxy groups. The conformation in Figure 1 generated by Glide with both methoxy groups in-plane and the one in Figure 3 from BOMB with one almost perpendicular to the benzene ring differ in energy by only 0.16 kcal/mol for o-dimethoxybenzene at the MP2/6-311G** level.23 Scheme 1. Recent NNRTI Hits from HTS (Ref 12)



Table 1. Anti-HIV-1 Activity (EC₅₀) and Cytotoxicity (CC₅₀) in μ M, BOMB Score, and Protein-Ligand Interaction Energy



					BOMB		
R_1	R_2	\mathbf{R}_3	R_4	EC_{50}^{a}	CC_{50}^{b}	score	$E_{\rm XX}^c$
CH ₃	Н	3-OCH ₃	4-OCH ₃	NA	61	-4.1	-59.3
Cl	Н	4-Cl	Н	NA	>100	-5.5	-56.1
Cl	Н	3-C1	Н	>10	32	-5.7	-53.7
Cl	Н	3-CH ₃	5-CH ₃	NA	>100	-5.7	-57.4
Cl	Н	2-C1	6-C1	0.82	20	-5.9	-61.3
Cl	Н	Н	Н	NA	31	-5.3	-55.6
Н	Н	Н	Н	NA	>100	-4.5	-49.8
Cl	CH ₃	2-C1	6-Cl	1.3	>100	-5.8	-64.6
2-Cl,4-Cl	OCH ₃	2-C1	6-Cl	NA	9	-5.4	-68.4
Cl	Cl	2-C1	6-Cl	0.31	>100	-5.7	-65.2
Cl	CF ₃	2-C1	6-Cl	NA	9	-6.2	-63.8
				1.6	>100		
				0.11	>10		
	R ₁ CH ₃ Cl Cl Cl Cl Cl 2-Cl,4-Cl Cl Cl Cl	$\begin{array}{ccc} R_1 & R_2 \\ \hline CH_3 & H \\ Cl & H \\ H & H \\ Cl & CH_3 \\ 2\text{-}Cl,4\text{-}Cl & OCH_3 \\ Cl & Cl \\ Cl & CF_3 \\ \hline \end{array}$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$				

^{*a*} For 50% protection in MT-2 cells; antiviral curves used triplicate samples at each concentration. NA for $EC_{50} > CC_{50}$. ^{*b*} For 50% inhibition of MT-2 cell growth; toxicity curves also used triplicate samples. ^{*c*} From conjugate gradient optimization of the complex using a dielectric constant of 2 with *MCPRO*; units are kcal/mol.

BOMB was then used to build the 100 complexes corresponding to the replacement of each phenyl hydrogen in the core with the 10 small groups. This revealed that the top-25 scoring analogs are dominated by substitution of either chlorine or a methyl group at the 3-, 4-, and 2-position in the benzyl ring and at the 4-, 2-, and 3-position of the anilinyl ring in roughly that order. The top three analogs have chlorine at the 3-, 4-, and 5-position in the benzyl ring, where the methoxy groups in Figure 3 would be said to be at the 3- and 4-positions. Monte Carlo/free-energy perturbation calculations in explicit water were also performed3 for mono-chlorination and favored substitution at the 2- and 6-positions of the benzyl ring and the 4-position of the anilinyl ring. The 45 possible dichlorosubstituted compounds were built next with BOMB. The 10 best scores are within 0.2 log unit, but revealed a preference for chlorine at the 4-position in the anilinyl ring and at the 5and 6-positions in the benzyl ring.

Given this information, synthesis of compounds 11-20 in Table 1 was initiated. The table includes the BOMB scores for 1 and the prepared compounds as well as the protein-ligand interaction energy for the complexes after conjugate-gradient optimization of the structures from BOMB using MCPRO²⁴ with the OPLS-AA (protein) and OPLS/CM1A (ligand) force fields and a dielectric constant of 2. At this point, it is noted that 1 scored poorly in comparison to the polychloro analogs; the difference was dominated by 1 being predicted to be less lipophilic by $1-2 \log$ units in QPlogP_{o/w} (2.6 vs 3.6-4.2) and for having 28 "bad" contacts versus ca. 18 for the dichloro compounds. The additional undesirable contacts stem from the proximity of the methoxy oxygens and the aromatic and C_{β} carbons of Tyr181 and Tyr188. Furthermore, a global concern for the series was the proximity of the carboxylate oxygens of Glu138 to N4 of the oxadiazole and to the benzylic CH₂ group.

Synthesis and Assay Results. The oxadiazole derivatives were prepared via cyclization of substituted phenylacetic hydrazides²⁵ and substituted phenyl isocyanide dichlorides,²⁶ as illustrated in Scheme 2 for 14. The procedure finds precedent in the synthesis of N,N-diphenyloxadiazol-2,5-diamines from semicarbazides.²⁷ Full synthetic and spectral details are provided in the Supporting Information. The assay results for the first three compounds that were prepared (11-13) were hardly encouraging. Compounds 11 and 13 are benign (inactive and non-cytotoxic), and for 12 there was, at best, a small hint of activity above 10 μ M that was masked by the onset of cytotoxicity with a CC50 of 32 µM. It was also somewhat surprising that, if 12 with a 3-chloro substituent in the benzyl ring were active, the 3,5-dimethyl analog (13) showed no activity up to 100 μ M. A 3,5-dimethylphenoxy group is present as the U group in the pyridinone ligand (a 2-nM NNRTI) in the 1s9e crystal structure,²² but it is clearly ineffective in the present oxadiazole series.

Nevertheless, **14** was then prepared to probe the 2,6-dichloro pattern that was supported by the computational analyses and has precedent in the 2,6-difluoro and 2,6-dimethyl substitution for the U groups in **2** and **4**. Compound **15** was also prepared as a reference compound, although there was no hope for it as an active in view of the results for the dichloro-analogs **11** and **12**. Fortunately, **14** was found to be a definite active at the 820-nM level and a respectable low-molecular-weight (355) lead for further optimization. Thus, **1** can be classified as a nearmiss with a need, at least, for an alternative substitution pattern in the benzyl ring. For reference, it is noted that the nucleoside inhibitor d4T and the NNRTI nevirapine are active at the 1.6- μ M and 110-nM levels in the MT-2 assay (Table 1).

LT00141083 Interlude. At this point, a paper appeared reporting the results of another screening effort seeking NNR-

Scheme 2. Synthesis of 2,5-Oxadiazole Derivatives



TIs.²⁸ In that case, a 3D-pharmacophore model was used for initial filtering of the Derwent and CAP databases, which was followed by docking of ca. 10 000 compounds with Glide. Six compounds were purchased to test their "in vitro RNAdependent DNA polymerase activity of recombinant RT." Five of the six compounds were reported to be active. This includes a compound 5 in ref 28 that was drawn as being identical to the core in the present work (16 in Table 1). No spectral or other characterization data were provided for the assayed compounds.²⁸ Compound 5 in ref 28 yielded an IC₅₀ of $4.2 \,\mu\text{M}$ in the enzyme assay; nevirapine was also reported as $0.18 \,\mu\text{M}$ for reference, which is a little less potent than that in our assay (Table 1). There was one available commercial source for 16, LaboTest AB, and the compound was designated LT00141083. We purchased it. The authors of the other study²⁸ verified that they had also purchased their compound 5 from LaboTest. A ¹H NMR spectrum was taken of the purchased sample, which immediately revealed that it could not be 16 in view of having the incorrect number of protons and, for example, no benzylic protons. Thus, 16 was then synthesized and submitted to the MT-2 anti-HIV assay. In view of the assay results for 11-15 and the BOMB scores, there was no possibility that 16 could be active. This was confirmed (Table 1). Under the circumstances, it seems highly unlikely that 16 was assayed in the other study.28 The 1H NMR spectra of the purchased material and 16 are compared in the Supporting Information.

A Few More Compounds. From the BOMB analyses and visualization of the resultant structures, replacement of hydrogen by chlorine or a methyl at C3 in the anilinyl ring of 14 seemed worth investigation. However, it is noted that such additions break the symmetry of the para-substituted phenyl ring. This can have negative impact on binding if the added substituent in the bound state can only be at either effectively the 3- or 5-position, while it can sample both in the unbound state. This could lead to a factor of 2 reduction in the binding affinity or perhaps even a factor of 10 in an extreme case in which the rotation of the phenyl ring is free in the unbound state, but restricted to a dihedral range of 36° when bound. For the present case, substitution of chlorine or a methyl group at C3 or C5 appeared to be allowed sterically and yielded similar BOMB scores, while a trifluoromethyl group is only allowed on the side facing the benzyl group in the bound state.

In the event, **17** and **19** were synthesized along with the trifluoromethyl analog **20**. An additional product, **18**, which arose in the course of failed attempts to synthesize the 3-methoxy analog, was also assayed. The methyl derivative **17** did not improve on the activity of **14**, and **18** and **20** were inactive, but the chloro analog **19** is a 310-nM NNRTI (Table 1). There is clearly room for additional lead optimization in this series, but it has been demonstrated that a near-miss from

virtual screening can be evolved into a true active. Finally, in comparing the experimental activities, BOMB scores, and protein–ligand interaction energies (E_{XX}) in Table 1, correlations are not perfect. However, the compounds with BOMB scores above -5.7 are all inactive. E_{XX} is not a particularly good index of potential binding because it tends to increase with increasing size of the ligand; dividing it by the number of atoms or molecular weight is flawed without correction for parts of the ligand that make no contact with the protein. Nevertheless, compounds with both very favorable BOMB scores and E_{XX} values are likely to be active. Compound **20** is the exception, although it suffers from a symmetry loss of at least a factor of 2 in activity.

Conclusions

Virtual screening of the Maybridge library of ca. 70 000 compounds was performed using a similarity filter, docking, and MM-GB/SA postprocessing to seek potential NNRTIs. The protocol did well in identifying known NNRTIs. Six representative, top-scoring compounds were purchased from Maybridge. Two were impure and gave incorrect mass spectra. After purification, the other four were submitted to an anti-HIV assay using infected human T-cells and found to show no ability to inhibit HIV replication. However, the highest-ranked library compound, the oxadiazole derivative 1, was subjected to further computational analysis to seek modifications that might yield an active anti-HIV agent. A small substituent scan was performed for the anilinylbenzyloxadiazole core with the BOMB program, which specifically suggested removal of the methoxy groups in the benzyl ring and the addition of two or three chlorines or methyl groups. Synthesis and assaying of several such compounds did vield active anti-HIV agents, as summarized in Table 1. The study demonstrates that it is possible to learn from a formally unsuccessful virtual-screening exercise and, with the aid of computational analyses, to efficiently evolve a false positive into a true active. The present case may be viewed as favorable in view of the availability of significant structure-activity data for NNRTIs; however, it provides a prototype for future efforts and, in fact, success was far from guaranteed owing to the sensitivity of activity to structure, the plethora of incorrect choices that could be made, and the lack of precedent for five-membered-ring heterocycles as the central ring in NNRTIs.

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Supporting Information Available: Synthetic details, NMR and HRMS spectral data for the oxadiazole derivatives in Table 1,

the ¹H NMR spectrum of LT00141083, and a figure illustrating the complex of **19** and HIV-RT. These materials are available free of charge via the Internet at http://pubs.acs.org.

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